Product Information



Gallic Acid

Item No. 11846

CAS Registry No.: 149-91-7

3,4,5-trihydroxy-benzoic acid Formal Name: Synonyms: NSC 20103, NSC 674319,

3,4,5-Trihydroxybenzoic acid

MF: $C_7H_6O_5$ FW: 170.1 **Purity:** ≥98%

Stability: ≥2 years at -20°C Supplied as: A crystalline solid UV/Vis.: λ_{max} : 218, 273 nm

COOH

Laboratory Procedures

For long term storage, we suggest that gallic acid be stored as supplied at -20°C. It should be stable for at least two years. Gallic acid is supplied as a crystalline solid. A stock solution may be made by dissolving the gallic acid in the solvent of choice. Gallic acid is soluble in organic solvents such as ethanol, DMSO, and dimethyl formamide (DMF), which should be purged with an inert gas. The solubility of gallic acid in ethanol and DMSO is approximately 16 mg/ml and approximately 25 mg/ml in DMF.

Further dilutions of the stock solution into aqueous buffers or isotonic saline should be made prior to performing biological experiments. Ensure that the residual amount of organic solvent is insignificant, since organic solvents may have physiological effects at low concentrations. Organic solvent-free aqueous solutions of gallic acid can be prepared by directly dissolving the crystalline solid in aqueous buffers. The solubility of gallic acid in PBS, pH 7.2, is approximately 2 mg/ml. We do not recommend storing the aqueous solution for more than one day.

Gallic acid is a trihydroxybenzoic acid found in many plants as either the free acid or in the esterified form of gallotannins and ellagitannins. It demonstrates antioxidant activity by scavenging 2,2-diphenyl-1-picrylhydrazyl and hydroxyl free radicals with IC₅₀ values of 9.4 and 191 μM, respectively, and inhibiting microsomal lipid peroxidation with an IC₅₀ value of 1.51 μM. Gallic acid is often used as a standard for determining the phenol content of various analytes by the Folin-Ciocalteau assay where results are reported in gallic acid equivalents.^{2,3}

- 1. Cos, P., Hermans, N., Calomme, M., et al. Comparative study of eight well-known polyphenolic antioxidants. J. Pharm. Pharmacol. 55, 1291-1297 (2003).
- 2. Bonilla, F., Mayen, M., Merida, J., et al. Extraction of phenolic compounds from red grape marc for use as food lipid antioxidants. Food Chem. 66, 209-215 (1999).
- Modun, D., Music, I., Vukovic, J., et al. The increase in human plasma antioxidant capacity after red wine consumption is due to both plasma urate and wine polyphenols. Atherosclerosis 197, 250-256 (2008).

Related Products

For a list of related products please visit: www.caymanchem.com/catalog/11846

WARNING: This product is for laboratory research only: not for administration to humans. Not for human or veterinary DIAGNOSTIC OR THERAPEUTIC USE.

This material should be considered hazardous until information to the contrary becomes available. Do not ingest, swallow, or inhale. Do not get in eyes, on skin, or on clothing. Wash thoroughly after handling. This information contains some, but not all, of the information required for the safe and proper use of this material. Before use, the user must review the complete Safety Data Sheet, which has been sent via email to your institution.

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