# **Product Information**



# TCEP (hydrochloride)

Item No. 14329

CAS Registry No.: 51805-45-9

Formal Name: 3,3',3"-phosphinylidynetris-propanoic

acid, monohydrochloride

Synonym: *Tris*(2-carboxyethyl)phosphine

MF:  $C_9H_{15}O_6P \cdot HCl$ 

FW: 286.7 **Purity:** ≥98%

Stability: ≥2 years at -20°C Supplied as: A crystalline solid

### **Laboratory Procedures**

For long term storage, we suggest that TCEP (hydrochloride) be stored as supplied at -20°C. It should be stable for at least two years.

TCEP (hydrochloride) is supplied as a crystalline solid. A stock solution may be made by dissolving the TCEP (hydrochloride) in the solvent of choice. TCEP (hydrochloride) is soluble in organic solvents such as DMSO and dimethyl formamide, which should be purged with an inert gas. The solubility of TCEP (hydrochloride) in these solvents is approximately 3.3 and 2 mg/ml, respectively.

Further dilutions of the stock solution into aqueous buffers or isotonic saline should be made prior to performing biological experiments. Ensure that the residual amount of organic solvent is insignificant, since organic solvents may have physiological effects at low concentrations. Organic solvent-free aqueous solutions of TCEP (hydrochloride) can be prepared by directly dissolving the crystalline solid in aqueous buffers. The solubility of TCEP (hydrochloride) in PBS, pH 7.2, is approximately 5 mg/ml. We do not recommend storing the aqueous solution for more than one day.

TCEP is an odorless, selective, and water-soluble reducing agent that is commonly used in many laboratory applications. It is commonly used to rapidly reduce protein and peptide disulfide bonds. TCEP can be combined with proteases to simultaneously reduce and digest proteins prior to mass spectrometry in order to dramatically increase sequence coverage. It has also been used to measure ascorbic acid and dehydroascorbic acid in biological samples.<sup>2,3</sup>

# References

- 1. Zhang, H.-M., McLoughlin, S.M., Frausto, S.D., et al. Simultaneous reduction and digestion of proteins with disulfide bonds for hydrogen/deuterium exchange monitored by mass spectrometry. Anal. Chem. 82(4), 1450-1454
- 2. Lykkesfeldt, J. Determination of ascorbic acid and dehydroascorbic acid in biological samples by high-performance liquid chromatography using subtraction methods: Reliable reduction with tris[2-carboxyethyl]phosphine hydrochloride. Anal. Biochem. 282(1), 89-93 (2000).
- Sato, Y., Uchiki, T., Iwama, M., et al. Determination of dehydroascorbic acid in mouse tissues and plasma by using tris(2-carboxyethyl)phosphine hydrochloride as reductant in metaphosphoric acid/ethylenediaminetetraacetic acid solution. Biol. Pharm. Bull. 33(3), 364-369 (2010).

# Related Products

For a list of related products please visit: www.caymanchem.com/catalog/14329

WARNING: This product is for laboratory research only: not for administration to humans. Not for human or veterinary DIAGNOSTIC OR THERAPEUTIC USE.

This material should be considered hazardous until information to the contrary becomes available. Do not ingest, swallow, or inhale. Do not get in eyes, on skin, or on clothing. Wash thoroughly after handling. This information contains some, but not all, of the information required for the safe and proper use of this material. Before use, the user must review the complete Safety Data Sheet, which has been sent via email to your institution.

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