

## Colorimetric Microplate Assay for Lipid Peroxidation

### Product No. FR 22

For Research Use Only

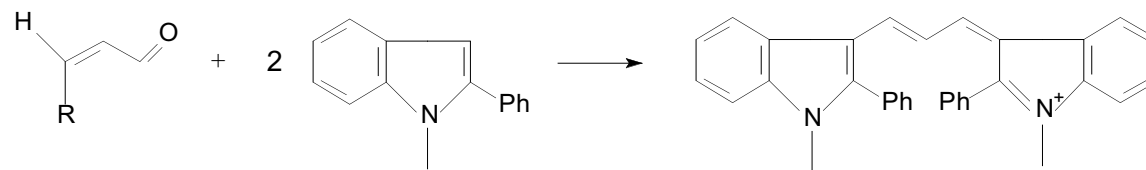
**WARNING:** This kit contains substances or reagents that may be harmful if used improperly. See MSDS for more information.

## INTRODUCTION

Lipid peroxidation, a well-established mechanism of cellular injury in plants and animals, is used as an indicator of oxidative stress in cells and tissues. Lipid peroxides are unstable and decompose to form a complex series of compounds including reactive carbonyl compounds. Polyunsaturated fatty acid peroxides generate malondialdehyde (MDA) and 4-hydroxyalkenals (HAE) upon decomposition, and the measurement of MDA and HAE has been used as an indicator of lipid peroxidation (1). This method is designed to assay either MDA alone (in hydrochloric acid) or MDA in combination with HAE (in methanesulfonic acid.)

## PRINCIPLES OF PROCEDURE

This assay is based on the reaction of a chromogenic reagent, N-methyl-2-phenylindole (R1), with MDA and 4-hydroxyalkenals at 45°C. One molecule of either MDA or 4-hydroxyalkenal reacts with 2 molecules of reagent R1 to yield a stable chromophore with maximal absorbance at 586 nm.



**MDA : R = OH**

**4-hydroxyalkenal : R= hydroxyalkyl**

**Max = 586 nm**

## MATERIALS PROVIDED (for 100 tests)

- |    |                     |  |
|----|---------------------|--|
| 1. | Reagent R1          | N-methyl-2-phenylindole in Acetonitrile, 1 x 13 mL           |
| 2. | Reagent R2          | Methanesulfonic acid (MSA), 2 x 2.2 mL <b>*Caustic Acid*</b> |
| 3. | MDA Standard        | 1,1,3,3-Tetramethoxypropane in Tris-HCl, 1 x 200 $\mu$ L     |
| 4. | Diluent             | Ferric Chloride Hexahydrate in Methanol, 2 x 2.6 mL          |
| 5. | 96 well microplate  | 12 strips x 8 wells in a frame                               |
| 6. | Microplate Template |  |
| 7. | Kit Instruction     |  |

## MATERIALS NEEDED BUT NOT PROVIDED

- Water bath or heat block set at 45°C
- 12N HCl, 37% (v/v)
- Butylated hydroxytoluene (BHT)
- Acetonitrile

5. Microcentrifuge
6. Microcentrifuge tubes (100 x 1.5 mL polypropylene tube or equivalent)
7. Microplate reader capable of measuring absorbance at 586 nm (a 580 nm or 590 nm filter will also provide acceptable results)

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## STORAGE

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Store this kit and its components at 4°C.

For information about partial kit storage, please see Procedural Notes #2.

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## WARNINGS AND PRECAUTIONS

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1. Use aseptic technique when opening and dispensing reagents.
2. This kit is designed to work properly as provided and instructed. Additions, deletions or substitutions to the procedure or reagents are not recommended, as it may be detrimental to the assay.
3. Wear proper protective eyewear and gloves when handling samples and reagents.
4. Do not pipette reagents or samples by mouth.
5. In case of accidental exposure of skin, mucous membranes or eyes to **R1** or **R2** reagents, thoroughly wash the exposed area with water and seek immediate medical attention.

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## PROCEDURAL NOTES

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1. Reagents can be used immediately upon removal from refrigeration.
2. When performing this kit in part, please adhere to the following:
  - All unused components should be returned to storage at 4 °C.
  - Unused portions of the microplate should be returned to the bag provided.
  - Create a standard curve for each performance of the assay.
3. Do not leave the reagent bottles open. Replace the caps as soon as the desired volume is removed.
4. Do not allow the capped reagent bottles to sit at room temperature for long periods of time. When not in use, place the bottles on ice or at 4°C.
5. To minimize error due to handling, wipe the exterior bottom of the microplate wells with a lint-free paper towel.
6. If reagents are handled and stored properly as described above, they are stable until the indicated expiration date.

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## SAMPLE COLLECTION AND STORAGE

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Unless assayed immediately, samples should be frozen at –70°C to prevent loss of MDA and (HAE) from *ex vivo* oxidation. Once thawed, samples should not be refrozen, and they should be protected from light to avoid photooxidation.

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## SAMPLE PREPARATION

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To prevent sample oxidation, it is recommended that BHT be added to a final concentration of 5 mM in the buffer prior to homogenization as instructed in Step 2 of the homogenization procedure below. BHT can be made as a 100x (500 mM) stock solution in acetonitrile. If no antioxidant is added, *ex vivo* lipid peroxidation can occur during homogenization and elevated values will result.

Sample homogenates should be prepared as concentrated as possible. The concentration of protein in the homogenate should be determined. It is recommended that 0.2 mL of a homogenate containing 15-69 mg/mL protein be assayed for initial studies in a previously untested biological sample. For tissue culture cells, it is recommended that at least  $1 \times 10^7$  cells per mL is used.

## PLASMA OR SERUM

The amount of free MDA or HAE in normal plasma or serum is at or below the limit of detection of this assay.

## CELL CULTURE AND TISSUE PREPARATION

Cell culture and tissue culture samples may require washing prior to homogenization.

**Cells** cultured in medium containing serum should be washed several times to remove serum components prior to homogenization. It is recommended that a homogenate of approximately  $5 \times 10^7$  cells per mL be prepared for use in the assay. Cells can be washed in ice-cold isotonic saline (i.e., 0.9% NaCl). The cells are now ready for the homogenization procedure below.

**Tissues** may require the removal of blood by perfusion *in situ* with isotonic saline or *in vitro* by rinsing with ice-cold isotonic saline and the perfusate discarded. After washing, the tissue should be weighed prior to the homogenization procedure below.

## HOMOGENIZATION PROCEDURE

1. Prepare homogenates in 20 mM phosphate buffer, pH 7.4. Other buffers may be used, but the researcher should confirm non-interference in the assay by measuring the MDA and/or 4-HNE standards diluted in the chosen buffer.
2. Prior to homogenization, add 10  $\mu$ L 0.5 M BHT in acetonitrile to 1 mL of sample to prevent sample oxidation. A precipitate is expected. It will be removed by centrifugation and will not affect the outcome of the assay.
3. Centrifuge the homogenate to remove large particles (e.g., 3000 x g at 4°C for 10 minutes).
4. Remove an aliquot of the sample for protein determination.
5. Freeze the sample immediately at -70°C or keep on ice prior to testing. If desired, aliquot the samples to provide 0.2 mL of the homogenate for each assay.

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## REAGENT PREPARATION

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Make the diluted R1 solution for the assay by adding one volume of Diluent to three volumes of Reagent R1. Prepare immediately before use. Do not leave the Reagent R1 bottle uncapped and open to the atmosphere.

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## STANDARD CURVE PREPARATION

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Malondialdehyde is provided as an acetal because the aldehyde is not stable. The acetal (TMOP) is hydrolyzed during the acid incubation step, which generates MDA.

**Note:** The table below shows how to create an eight point standard curve using 24 wells. Fewer concentrations may be used at the discretion of the investigator.

The TMOP standard is provided as a 10 mM stock solution. Dilute the 10 mM stock TMOP 1:500 (v/v) in water just prior to use yielding a 20  $\mu$ M stock solution. For a standard curve, combine the volumes shown in Table 1 for 20  $\mu$ M stock solution and H<sub>2</sub>O to give a total of 180  $\mu$ L for each respective standard. This schematic will provide sufficient volume for assay in triplicate.

**Table 1.**

Standard #	S <sub>1</sub>	S <sub>2</sub>	S <sub>3</sub>	S <sub>4</sub>	S <sub>5</sub>	S <sub>6</sub>	S <sub>7</sub>	S <sub>8</sub>
Concentration ( $\mu$ M)	0	0.1	0.2	0.5	1	2	3	4
20 $\mu$ M Stock ( $\mu$ L)	0	5	9	22	45	90	135	180
H <sub>2</sub> O ( $\mu$ L)	180	175	171	158	135	90	45	0

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## ASSAY PROCEDURE

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**NOTE:** Centrifugation is absolutely necessary due to the presence of a precipitate in solution. Be careful not to disturb this pellet after centrifugation, as it will interfere with accurate readings. If the pellet is disturbed, centrifuge the sample again.

**Reagent Blank:** Create a reagent blank ( $B_0$ ) to correct for any contribution to the  $A_{586}$  due to the chromophore in the sample. A 325  $\mu\text{L}$  blank is made by combining one volume of the provided Diluent with three volumes of acetonitrile. Use this blank in place of the diluted Reagent R1 as described in step 2 of the assay procedures below.

**Sample Blank:** A sample blank (water substituting for sample) is included in the standard curve preparation ( $S_1$ ).

1. Add 140  $\mu\text{L}$  of sample, standard, or blank to a clean, labeled polypropylene microcentrifuge tube.
2. Add 455  $\mu\text{L}$  of diluted Reagent R1 (see Reagent Preparation) to each tube.
3. Mix gently by vortexing.
4. **NOTE: For MDA only:** Add 105  $\mu\text{L}$  of 12N HCl (37% HCl) to each tube.  
**For MDA + HAE:** Add 105  $\mu\text{L}$  of Reagent R2 to each tube.
5. Mix well and close cap.
6. Incubate at 45°C for 60 minutes.
7. Centrifuge turbid samples to obtain clear supernatant (13,000 x g for 15 minutes).
8. Transfer 3 x 150  $\mu\text{L}$  of the clear supernatant to a microplate (see template below).
9. Measure absorbance at 586 nm. If this is not available, a 580 nm or 590 nm filter can be used. The color is stable for at least one hour at room temperature or 2 hours at 4°C if the plate is stored in the dark.

#### Scheme I

	1	2	3	4	5	6	7	8	9	10	11	12
A	S1	S1	S1	B0	B0	B0	U8	U8	U8	U16	U16	U16
B	S2	S2	S2	U1	U1	U1	U9	U9	U9	U17	U17	U17
C	S3	S3	S3	U2	U2	U2	U10	U10	U10	U18	U18	U18
D	S4	S4	S4	U3	U3	U3	U11	U11	U11	U19	U19	U19
E	S5	S5	S5	U4	U4	U4	U12	U12	U12	U20	U20	U20
F	S6	S6	S6	U5	U5	U5	U13	U13	U13	U21	U21	U21
G	S7	S7	S7	U6	U6	U6	U14	U14	U14	U22	U22	U22
H	S8	S8	S8	U7	U7	U7	U15	U15	U15	U23	U23	U23

#### CALCULATIONS

In HCl and MSA, the molar extinction coefficient ( $\epsilon$ ) at 586 nm for MDA is approximately 120,000. The color yield is a linear function of the MDA concentration over the range from 0  $\mu\text{M}$  to 20  $\mu\text{M}$ . 4-HNE cannot be measured in the HCl protocol, as it does not form a chromophore with the reagent in the presence of HCl. For simultaneous determination of both MDA and 4-HNE, the MSA procedure must be used. The extinction coefficient using MSA is determined empirically from the slope of a standard curve in a manner similar to that described in the Example.

1. Using the standard data, calculate the net  $A_{586}$  for each standard by subtracting the blank ( $A_0$ ) value from each of the standard  $A_{586}$  values. Plot net  $A_{586}$  vs. [MDA], and perform a linear regression analysis of  $A_{586}$  on [MDA]:

$$[MDA] = a[A_{586}] + b$$

2. Calculate the concentration of analyte in each unknown from the net  $A_{586}$  of the sample (If a sample blank was required, subtract  $A_{sb}$  from the net sample absorbance.):

$$[MDA] = \frac{A_{586} - b}{a} \cdot df$$

Where:

[MDA] is the  $\mu\text{M}$  concentration of MDA in the sample

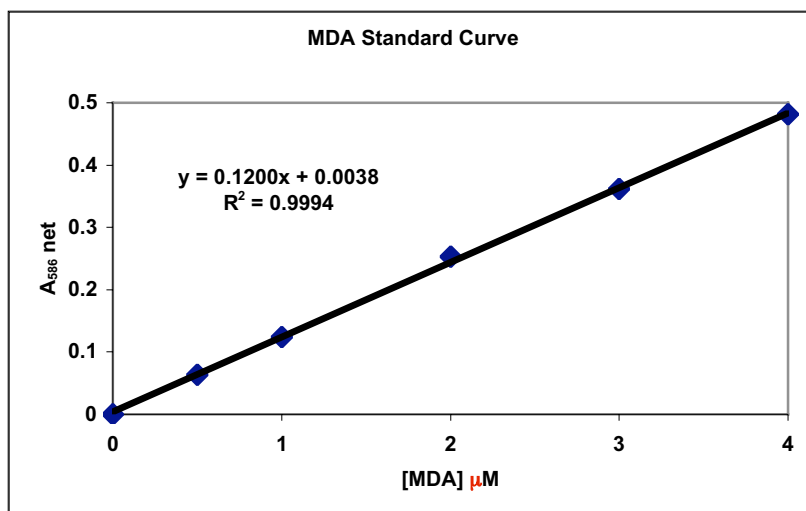
$A_{586}$  = Net absorbance at 586 nm of the sample

a = regression coefficient (slope)

b = intercept

df = dilution factor

**Figure 1: Typical Standard Curve**



### Example:

To illustrate the appropriate calculations, consider the following experiment:

Three 200 mL aliquots of a biological sample were assayed along with a triplicate set of MDA standards. A sample blank had an absorbance of 0 at 586 nm. The average  $A_{586}$  value of the zero concentration standard was subtracted from the average  $A_{586}$  values of the other standards and the sample to give corrected absorbencies ( $A_{586}$  net). A plot of  $A_{586}$  net vs. [MDA] for the standards was constructed (see Figure 1 above).

The sample had an average  $A_{586}$  net = 0.2752. Steps in the calculation of sample [MDA] are:

1. Calculate the [MDA] in the reaction mixture:

$$[\text{MDA}] = \frac{0.2752 - 0.0038}{0.1200} = 2.261 \mu\text{M}$$

2. Correct for the dilution factor of the sample:

140 mL of sample was used and the final reaction volume was 700 mL.

$$\text{df} = \frac{700}{140} = 5.000$$

$$[\text{MDA}] \text{ in biological sample} = 2.261 \text{ mM} \times 5.000 = 11.3 \text{ mM}$$

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## PERFORMANCE CHARACTERISTICS

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### LEAST DETECTABLE CONCENTRATION

Experiments employing enaldehyde standards and blanks suggest that the analytical limit of detection in a purified system is 0.1 nmol/mL final concentration (0.5 nmol/mL in the sample), corresponding to an absorbance value of approximately 0.011. The limit for biological samples will generally be higher. It is recommended that the researcher confirm this parameter in their system.

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**REPRODUCIBILITY**

Experiments in which standard samples (0-20  $\mu$ M) were assayed using the same protocol over a period of 10 days established that the standard error of the measurement (SEM) is less than 5%.

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**REFERENCES**

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